## A CONVENIENT ONE-POT SYNTHESIS OF α-HALOACYLSILANES

Philip C. Bulman Page\* and Stephen Rosenthal,

Department of Organic Chemistry, Robert Robinson Laboratories, University of Liverpool, P.O.Box 147, Liverpool, L69 3BX, U.K.

Abstract:  $\alpha$ -Bromoacylsilanes may be prepared from trimethylsilylacetylenes by hydroboration-oxidation followed by treatment of the intermediate borinate ester with N-bromosuccinimide.

During the course of our studies of the chemistry of acylsilanes<sup>1,2</sup> we discovered a need for a simple and general preparative method for  $\alpha$ -haloacylsilanes (1).

$$X \longrightarrow O$$

$$R \longrightarrow SiR'_3$$

$$X = Br, CI$$

(1)

 $\alpha$ -Haloacylsilanes have previously been synthesised <u>via</u> a number of different routes including bromination of a 1,1-bis-trimethylsilylalkan-1-ol<sup>3</sup> and bromination of the silylenolether of an acylsilane;<sup>4,5</sup> however, these methods suffer from lengthy multi-step sequences, variable yields, or experimental difficulties.

We are, therefore, pleased to report a convenient synthesis of  $\alpha$ -bromoacylsilanes from trimethylsilylacetylenes using hydroboration methodology. It has been reported that enol borinates of simple aldehydes and ketones react with N-bromosuccinimide at 0°C to give the corresponding  $\alpha$ -brominated species. We have found that enol borinates of acyl silanes, prepared directly from silyl acetylenes  $\underline{via}$  hydroboration-oxidation, react in a similar manner to provide  $\alpha$ -bromoacylsilanes in moderate to good yields.

$$R = SiMe_{3} = \frac{1) BH_{3}.Me_{2}S}{2) Me_{3}NO} = R$$

$$R$$

$$R$$

$$CH_{3}(CH_{2})_{4} \qquad 61\%$$

$$CH_{3}(CH_{2})_{3} \qquad 60\%$$

$$CI(CH_{2})_{3} \qquad 61\%$$

$$^{\dagger}B_{0}Me_{2}SiO(CH_{2})_{3} \qquad 54\%$$

$$^{\dagger}B_{0}Me_{2}SiO(CH_{2})_{2} \qquad 52\%$$

The entire sequence may be carried out in one pot as described below.

## 2-Bromohexanoyl trimethylsilane

A 2 molar solution of borane-dimethylsulphide complex in THF (7.3 mmol) was added dropwise to a solution of 1-trimethylsilylhex-1-yne (3.08g, 20 mmol) in THF (5 ml) under a nitrogen atmosphere while maintaining the temperature below 20°C. The resultant solution of trivinylborane was diluted with THF (100 ml), treated with anhydrous trimethylamine-N-oxide (2.67g, 24 mmol) at room temperature over a five minute period, and refluxed under a nitrogen atmosphere for four hours. The slurry was cooled to 0°C and purified N-bromosuccinimide (4.27 g, 24 mmol) added. After stirring at 0°C for between one and four hours the solvent was removed in vacuo and a saturated aqueous solution of ammonium chloride (100 ml) added. The aqueous layer was extracted with hexane (2 x 100 ml) and the combined organic solutions washed with water (2 x 30 ml) and dried over magnesium sulphate. Removal of the solvent and purification by column chromatography gave 2-bromohexanoyl trimethylsilane (3.0g, 60%).

We thank the SERC for financial support.

## REFERENCES

- 1. P.C.B.Page and S.Rosenthal, Tetrahedron Lett., 1986, 27, 1947.
- 2. P.C.B.Page and S.Rosenthal, Tetrahedron Lett., 1986, 27, 2527.
- 3. I.Kuwajima, T.Abe, and N.Minami, Chem. Lett., 1976, 993.
- 4. I. Kuwajima, T. Sato and T. Abe, Tetrahedron Lett., 1978, 259.
- 5. T.Cohen and J.R.Matz, J.Am. Chem. Soc., 1980, 102, 6900.
- 6. J.A.Miller and G.Zweifel, Synthesis, 1981, 288.
- 7. J. Hooz and J.N. Bridson, Can. J. Chem., 1972, 50, 2387.

(Received in UK 3 September 1986)